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SAFETY DATA SHEET

1. IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product Name

Zephex™ 134a

REACH Registration No.

01-2119459374-33-0000

Manufacturer

Mexichem UK Limited

The Heath Business & Technical Park

Runcorn Cheshire WA7 4QX United Kingdom

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Emergency Phone No.

IN AN EMERGENCY DIAL 999 (UK Only)

For specialist advice in an emergency telephone +44(0) 1928 572000

Use

Subject to Member State regulations, applicable uses are: medical

propellant

24 = 15/4/2/13/2/5/10/3/10/13/13/13/0/4/13/0/A

Low acute toxicity. High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects and asphyxiation. Liquid splashes or spray may cause freeze burns to skin and eyes.

EC Classification

Regulation (EC) No. 1272/2008 (CLP)

Gases under pressure - Liquefied gas

Label elements

Hazard Statement(s)

H280: Contains gas under pressure; may explode if heated.

Signal Word(s)

Warning

Hazard Pictogram(s)



Precautionary Statement(s)

P410+P403: Protect from sunlight. Store in a well-ventilated place.

3. COMPOSITION / INFORMATION ON INGREDIENTS

Alternative names

HFA 134a

1,1,1,2-tetrafluoroethane (HFC 134a)

HAZARDOUS INGREDIENT(S)

Product Name Zephex™ 134a Revision: GHS03 Date: 06/2016 Page: 1 of 6



Hazardous Ingredient(s)	%(w/w)	CAS No.	EC No.	Hazard symbol(s) and hazard statement(s)
1,1,1,2-tetrafluoroethane (HFC 134a)	100	000811-97-2	212-377-0	GHS04 H280

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The first aid advice given for skin contact, eye contact, and ingestion is applicable following exposures to the liquid or spray. See also section 11.

Inhalation

Remove patient from exposure, keep warm and at rest. Administer oxygen if necessary. Apply artificial respiration if breathing has ceased or shows signs of failing. In the event of cardiac arrest apply external cardiac massage. Obtain immediate medical attention.

Skin Contact

Thaw affected areas with water. Remove contaminated clothing. Caution: clothing may adhere to the skin in the case of freeze burns. After contact with skin, wash immediately with plenty of warm water. If irritation or blistering occur obtain medical attention.

Eye Contact

Immediately irrigate with eyewash solution or clean water, holding the eyelids apart, for at least 10 minutes. Obtain immediate medical attention.

Ingestion

Unlikely route of exposure. Do not induce vomiting. Provided the patient is conscious, wash out mouth with water and give 200-300 ml (half a pint)

of water to drink. Obtain immediate medical attention.

Further Medical Treatment

Symptomatic treatment and supportive therapy as indicated. Adrenaline and similar sympathomimetic drugs should be avoided following exposure as cardiac arrhythmia may result with possible subsequent cardiac arrest.

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General

HFC 134a is not flammable in air under ambient conditions of temperature and pressure. Certain mixtures of HFC 134a and air when under pressure may be flammable. Mixtures of HFC 134a and air under pressure should be avoided.

Certain mixtures of HFCs and chlorine may be flammable or reactive under certain conditions. Thermal decomposition will evolve very toxic and corrosive vapours. (hydrogen fluoride)

Containers may burst if overheated.

Extinguishing media

As appropriate for surrounding fire.

Keep fire exposed containers cool by spraying with water.

Fire Fighting Protective Equipment

A self contained breathing apparatus and full protective clothing must be

worn in fire conditions. See Also Section 8

Personal Protection

Ensure suitable personal protection (including respiratory protection) during removal of spillages. See Also Section 8

General

Provided it is safe to do so, isolate the source of the leak. Allow small spillages to evaporate provided there is adequate ventilation. Large spillages: Ventilate area. Contain spillages with sand, earth or any suitable adsorbent material. Prevent liquid from entering drains, sewers, basements and workpits since the vapour may create a suffocating atmosphere.

Product Name Zephex™ 134a Revision: GHS03

Date: 06/2016 Page: 2 of 6



7 HANDLING AND STORAGE

Handling

Avoid inhalation of high concentrations of vapours. Atmospheric levels should be controlled in compliance with the occupational exposure limit. Atmospheric concentrations well below the occupational exposure limit can be achieved by good occupational hygiene practice.

The vapour is heavier than air, high concentrations may be produced at low levels where general ventilation is poor, in such cases provide adequate ventilation or wear suitable respiratory protective equipment with positive air supply.

Avoid contact with naked flames and hot surfaces as corrosive and very toxic decomposition products can be formed.

Avoid contact between the liquid and skin and eyes.

Avoid venting to atmosphere.

The fluorinated greenhouse gas HFA 134a may be supplied in returnable containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in containers may not be vented to the atmosphere. Regulation (EC) No. 842/2006 of the European Parliament and the Council on certain fluorinated greenhouse gases.

Process Hazards

The transfer of liquid HFC134a between containers and to and from processing equipment can result in static generation. Ensure adequate partition

earthing

Care must be taken to mitigate the risk of developing high pressures in systems caused by a temperature rise when liquid is trapped between closed valves or in cases where containers have been overfilled.

Storage

Keep in a well ventilated place away from fire risk and avoid sources of

heat such as electric or steam radiators.

Avoid storing near to the intake of air conditioning units, boiler units and

open drains.

Specific use

Subject to Member State regulations, applicable uses are: medical

propellant

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

General

Wear suitable protective clothing, gloves and eye/face protection. Wear thermal insulating gloves when handling liquefied gases.

In cases of insufficient ventilation, where exposure to high concentrations of vapour is possible, suitable respiratory protective equipment with

positive air supply should be used.



Eye Protection



Gloves

Occupational Exposure Limits

Occupational Exposure Limits	CAS No.	LTEL (8 hr TWA ppm)	LTEL 8 hr TWA mg/m³	STEL (ppm)	STEL mg/m³	Note
1,1,1,2-Tetrafluoroethane (HFC 134a)	000811-97-2	1000	4240	-	-	WEL





YSIGALAND CHEWIGAL PROPERTIES

Form liquefied gas Colour. colourless slight ethereal Odour Solubility (Water) Solubility (Other) slightly soluble

Soluble in: alcohols, chlorinated solvents, esters, polyethylene glycol

Boiling Point (° C) -26.2 Melting Point (° C) -101

Vapour Density (Air=1) 3.66 at normal boiling point

Vapour Pressure (mm Hg) 4270 at 20 ° C Specific Gravity 1.22 at 20 ° C

ANNO ELEMANTAN

Certain mixtures of HFCs and chlorine may be flammable or reactive Hazardous Reactions

under certain conditions.

Incompatible materials: finely divided metals, magnesium and alloys containing more than 2% magnesium. Can react violently if in contact with alkali metals and alkaline earth metals - sodium, potassium, barium

Hazardous Decomposition Product(s) hydrogen fluoride by thermal decomposition and hydrolysis.

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Inhalation LC50 (rat) (4 hrs) > 500000 ppm (2080000 mg/m3)

High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause

anaesthetic effects and asphyxiation.

Skin Contact Liquid splashes or spray may cause freeze burns. Unlikely to be

hazardous by skin absorption.

Eye Contact Liquid splashes or spray may cause freeze burns.

Ingestion Highly unlikely - but should this occur freeze burns will result.

A lifetime inhalation study in rats has shown that exposure to 50000ppm Long Term Exposure

resulted in benign tumours of the testis. The increased tumour incidence was observed only after prolonged exposure to high levels, and is considered not to be of relevance to humans occupationally exposed to

HFC 134a at or below the occupational exposure limit.

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High tonnage material produced in wholly contained systems. High **Environmental Fate and Distribution**

tonnage material used in open systems. Vapour

Decomposed comparatively rapidly in the lower atmosphere Persistence and Degradation (troposphere). Atmospheric lifetime is 14 years. Products of

decomposition will be highly dispersed and hence will have a very low

concentration. Does not influence photochemical smog (i.e. is not a VOC

under the terms of the UNECE agreement).

Does not deplete ozone.

Has a Global Warming Potential (GWP) of 1300 (relative to a value of 1 for carbon dioxide at 100 years) according to Annex I of Regulation 842/2006 on certain fluorinated greenhouse gases. Values in Annex I are taken from the third assessment report (TAR) of the Intergovernmental

Panel on Climate Change (2001 IPCC GWP values).

United Nations Framework Convention on Climate Change (UNFCCC)

reporting GWP is 1300.

Date: 06/2016 Product Name Zephex™ 134a Revision: GHS03 Page: 4 of 6



Effect on Effluent Treatment

Discharges of the product will enter the atmosphere and will not result in long term aqueous contamination.

TREE DISPOSANCIONSIDERATIONS

Recommended:

Best to recover and recycle. If this is not possible, destruction is to be in an approved facility which is equipped to absorb and neutralise acid gases and other toxic processing products.

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Hazard label(s)



Road/Rail

UN No.

ADR/RID Class

ADR/RID Proper Shipping Name

3159 2.2

1,1,1,2-TETRAFLUOROETHANE (REFRIGERANT GAS R 134a)

SEA

IMDG Class

Marine Pollutant

2.2

Not classified as a Marine Pollutant

AIR

ICAO/IATA Class

2.2

ETA MELACHIMATOR VALUE OF MATERIAL

European Regulations

EC Classification

Regulation (EC) No. 1272/2008 (CLP) Gases under pressure - Liquefied gas

Special Restrictions:

The fluorinated greenhouse gas HFA 134a may be supplied in returnable containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in containers may not be vented to the atmosphere.

Regulation (EC) No. 842/2006 of the European Parliament and the Council on certain fluorinated greenhouse gases.

46 - OTHERNIERS///488(0)\

This data sheet was prepared in accordance with Regulation (EC) No. 1907/2006.

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Product Name Zephex[™] 134a Revision: GHS03 Date: 06/2016 Page: 5 of 6



Glossary

WEL: Workplace Exposure Limit (UK HSE EH40)
COM: The company aims to control exposure in its workplace to this limit The company aims to control exposure in its workplace to the ACGIH limit TLV-C: The company aims to control exposure in its workplace to the ACGIH Ceiling limit MAK: The company aims to control exposure in its workplace to the German limit

Can be absorbed through skin

Capable of causing respiratory sensitisation

Bmgv: Biological monitoring guidance value (UK HSE EH40)

Hazard Statement(s)

H280: Contains gas under pressure; may explode if heated.

The following sections contain revisions or new statements: 1,2,15,16

Product Name Zephex™ 134a Revision: GHS03

Date: 06/2016 Page: 6 of 6